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Interactions between polycyclic aromatic hydrocarbons in complex mixtures

and implications for cancer risk assessment

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Abbreviations

AhR, Aryl hydrocarbon receptor; ATM, ataxia telangiectasia mutated; ATR, ataxia telangiectasia and Rad3-related protein; B[a]P, benzo[a]pyrene; B[a]P_{eq}, B[a]P-equivalent; Chk1, checkpoint kinase 1; CT, coal tar; CYP, cytochrome P450; DE, diol epoxide; DNA-PK, DNA-dependent protein kinase; DPE, diesel particulate extract; DPM, diesel particulate matter; EFSA, European Food Safety Agency; H2AX, H2A histone family member X; HEL, human embryonic lung; IARC, International Agency for Research on Cancer; MAF, mixture assessment factor; NTP, National Toxicology Program; PAH, polycyclic aromatic hydrocarbon; PCB, polychlorinated biphenyl; PCDD, polychlorinated dibenzo-p-dioxin; PCDF, polychlorinated dibenzofuran; PCN, polychlorinated naphthalene; PEF, potency equivalency factor; ROS, reactive oxygen species; RPF, relative potency factor; SRM, standard reference material; TBA, tumor bearing animal; TEF, toxic equivalency factor; UD, urban dust;

Abstract

In this review we discuss the effects of exposure to complex PAH mixtures in vitro and in vivo on

mechanisms related to carcinogenesis. Of particular concern regarding exposure to complex PAH

mixtures is how interactions between different constituents can affect the carcinogenic response and

how these might be included in risk assessment. Overall the findings suggest that the responses

resulting from exposure to complex PAH mixtures is varied and complicated. More- and less-than

additive effects on bioactivation and DNA damage formation have been observed depending on the

mixtures studied, and equally dependent on the different test systems that are used. Furthermore, the

findings show that the commonly used biological end-point of DNA damage formation is

insufficient for studying mixture effects. At present the assessment of the risk of exposure to

complex PAH mixtures involves comparison to individual compounds using either a surrogate or a

component-based potency approach. We discuss how future risk assessment strategies for complex

PAH mixtures should be based around whole mixture assessment in order to account for interaction

effects. Inherent to this is the need to incorporate different experimental approaches using robust

and sensitive biological endpoints. Furthermore, the emphasis on future research should be placed

on studying real life mixtures that better represent the complex PAH mixtures that humans are

exposed to.

Key Words

Benzo[a]pyrene; Cancer; Complex Mixtures; Polycyclic Aromatic Hydrocarbons; Risk Assessment

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1. Introduction

The field of mixture toxicology and the experimental analysis of chemical mixtures have undergone a significant expansion borne out of the need to understand the effects of exposure on human health. Historically, experimental approaches have focused on understanding the effects of individual or simple combinations of chemicals. However, this approach is being superseded in favor of investigating the effects of complex mixtures and understanding of the interactions between different chemicals. A review published in the early 1990's summarized the then status of toxicological assessment of complex mixture effects (Mauderly 1993). That the issues pertaining to complex mixture assessment presented at that time are still as prevalent today highlights the challenges faced by researchers in this field.

Humans are exposed to complex mixtures of chemicals through various sources including occupational settings, the environment, cigarette smoking, vehicular exhaust emissions, pharmaceuticals, lifestyle products and foodstuffs. Exposure occurs over a prolonged period of time and generally at low levels. Typically this exposure includes a vast assortment of chemicals including, but not limited to, polycyclic aromatic hydrocarbons (PAHs), polychlorinated biphenyls (PCBs), dibenzo-p-dioxins (PCDDs), dibenzofurans (PCDFs) and naphthalenes (PCNs), pesticides and heavy metals. There is however a significant lack of information available today regarding complex mixture toxicology.

A major group of chemicals that are found in complex mixtures are the PAHs, a family of more than 1500 compounds (NTP 2012) comprised of two or more fused aromatic rings, found both in their native and substituted forms (i.e. methylated, oxygenated or nitrated) (Figure 1). PAHs are ubiquitous environmental pollutants that are formed as a result of incomplete pyrolytic processes and to which humans are exposed through inhalation, ingestion and dermal absorption. Links

between human exposure to complex PAH mixtures and development of diseases including cancer, and respiratory and cardiovascular diseases have been described previously (ATSDR 1995; Boffetta et al. 1997; IARC 2010; Peters et al. 2001; Pope et al. 2002). Despite their structural similarities, PAHs vary greatly in their carcinogenic potency, with both individual and complex mixtures of PAHs classified as possible or probable human carcinogens by the International Agency for Cancer Research (IARC) (IARC 2010).

Assessment of the carcinogenic risk to humans of exposure to PAHs often involves comparison to benzo[a]pyrene (B[a]P) and two approaches are commonly used (Boström et al. 2002; Pufulete et al. 2004). In the first approach B[a]P is used as a surrogate marker to determine quantitative risk estimates for mixtures of PAHs in air (WHO 2000) or food (EFSA 2008). The second approach is component-based in which the potency of different PAHs is expressed relative to that of B[a]P (assigned a nominal value of 1). The assigned values for the different PAHs are termed Relative Potency Factors (RPF) or Toxic Equivalency Factors (TEF)/Potency Equivalent Factors (PEF). Detailed description of the differences between how the factors are derived have been published previously (Backhaus et al. 2010; U.S.EPA 2000). An important issue regarding the componentbased approach is that due to a lack of data not all PAHs have been assigned potency values and this is likely to lead to a misestimating of the risk. Principles pertaining to how risk assessments could be applied to complex PAH mixtures have been proposed (ATSDR 2004; EC 2009; Flowers et al. 2002; IPCS 2009; NTP 2012; SCHER et al. 2012; U.S.EPA 2007; U.S.EPA 2010), and indeed a recent research proposal testing the health effects of complex PAH mixtures has been approved (NTP 2012). However, it is accepted that due a number of limiting factors (discussed in detail below) this is likely to lead to misestimating of the risk to human health.

The aim of this paper is to review what is currently known about the effects of interactions between PAHs in complex mixtures on mechanisms related to carcinogenesis. We then discuss ways and

advantages of studying complex PAH mixtures and how such findings could be applied to future risk assessment.

2. Carcinogenic effects of PAHs

The binding of PAHs to DNA and the associated effects that occur as a result is considered the major mechanism of PAH-induced mutagenesis and carcinogenesis. Like many chemical carcinogens, PAHs require activation through a series of enzymatically-catalyzed reactions to form their active metabolites (Conney 1982; Huberman et al. 1976; Sims et al. 1974). The CYP family of enzymes, in particular CYP1A1, 1A2 and 1B1, are primarily involved in bioactivation of PAHs to their reactive intermediates (Pelkonen and Nebert 1982; Shimada and Fujii-Kuriyama 2004; Shimada et al. 1996). Many PAHs are ligands for the aryl hydrocarbon receptor (AhR), which has different roles involved in metabolism including regulation of the different bioactivating and detoxifying enzymes (Baird et al. 2005; Nebert et al. 2004). For example, B[a]P is activated through a three-step enzymatic mechanism involving initial metabolism by CYP enzymes to B[a]P-7,8-epoxides, followed by conversion to B[a]P-7,8-diols by epoxide hydrolase, and final transformation to the ultimate reactive B[a]P-7,8-diol-9,10-epoxide metabolites, again by CYP enzymes (Conney 1982). This mechanism of activation has also been shown for other PAHs. An additional metabolic pathway is the aldo-keto reductase mechanism that activates PAHs to redoxactive o-quinone derivatives which might also have tumorigenic and mutagenic activities (Park et al. 2008; Penning et al. 1999; Zhang et al. 2012). Although many substituted PAHs are also targets for CYPs, their bioactivation often involves additional steps. For example, nitrated PAHs (such as 1-nitropyrene) require initial nitroreduction to primary amines before they are metabolized by CYPs to hydroxylamines and activated by conjugation reactions (Chou and Fu 1983; Djuric et al. 1986).

The tumorigenic and mutagenic activities of many PAHs have been linked to the ability of their diol epoxide (DE) metabolites to bind covalently to exocyclic amino groups on purine bases to form either stable bulky PAH-DE-DNA adducts or depurinating adducts which are released from DNA leaving abasic sites (Rogan et al. 1993; Sims and Grover 1974; Szeliga and Dipple 1998). Quinones can also bind DNA and form stable N2-dG or N2-dA adducts (Penning et al. 1999; Shou et al. 1993) or depurinating N7-dG adducts (McCoull et al. 1999). PAHs exhibit a large variety in their ability to form adducts with DNA which may be a result of different structural conformations. Most carcinogenic PAHs contain either a bay or a fjord region within their structure (Figure 1). PAHs containing a bay region (i.e. B[a]P) are rigid and planar and their DEs typically react with guanine residues, whereas those containing a fjord region (i.e. dibenzo[a,l]pyrene, also known as dibenzo[def,p]chrysene) are sterically hindered and non-planar and typically react with adenine residues. Studies comparing the biological activity of bay- and fjord-region containing PAHs have shown that the latter in general have a higher mutagenic and tumorigenic potential (Dipple et al. 1987; Jerina et al. 1986; Ralston et al. 1995; Szeliga and Dipple 1998). The variation in potencies is most probably associated with the structural differences between adducts and the subsequent effects on removal by DNA repair mechanisms (Buterin et al. 2000; Dreij et al. 2005; Kropachev et al. 2013; Suh et al. 1995). However, it could also be a result of changes in DNA polymerase activity and incorrect base-pair insertion resulting from post-lesion DNA synthesis (Broyde et al. 2010; Eoff et al. 2010; Huang et al. 2003).

Following bioactivation and reaction with DNA, the PAH metabolites can induce mutations that either activate oncogenes or inactivate tumor suppressor genes as part of their carcinogenic mechanism. This process can be exacerbated in complex PAH mixtures as they might contain compounds that alone have weak carcinogenic potential (e.g. promoters) but in combination with initiators act in an additive or more-than additive manner (Baird et al. 2005; U.S.EPA 2007). If mutations occur in tumour suppressors this can lead to uncontrolled cell cycle regulation,

accumulation of DNA damage and ultimately carcinogenesis. For example, the *p53* tumor suppressor gene is mutated in many human cancers (Hainaut et al. 1998; Hollstein et al. 1991) and exposure to PAHs has previously been associated with mutations in p53 in human lung and breast cancers (Hussain et al. 2001; Mordukhovich et al. 2010). The *Ras* proto-oncogene is mutated in a large number of human cancers (Bos 1989; Prior et al. 2012) and associations with PAH exposure have been observed *in vivo* (Gray et al. 2001; Ross and Nesnow 1999).

In addition to the above described mechanisms, generation of reactive oxygen species (ROS) and oxidative DNA damage can occur. Induction of oxidative stress is a major mode of toxicity associated with airborne particles to which PAHs can be absorbed. This includes responses such as formation of oxidative DNA damage, triggering of an inflammatory response and cell death (Lodovici and Bigagli 2011). Oxidative DNA damage can result from PAHs undergoing rounds of futile redox cycling leading to the formation of 8-oxo-dGuo adducts with subsequent induction of mutations (Park et al. 2005; Park et al. 2006). However, it is often very hard to ascertain the contribution of oxidative DNA damages on PAH-induced carcinogenesis, in part because such damages are formed in parallel with DNA adducts, and thus, to date, the role of oxidative stress in PAH mixture toxicology remains unclear.

Pro-carcinogenic epigenetic mechanisms are evidenced as heritable changes that do not involve direct modification of the DNA sequence (Esteller 2008; Jones and Baylin 2007). Epigenetic changes including DNA methylation and telomere dysfunction have been reported after exposure to complex PAH mixtures in human workers and following *in vivo* mouse exposure (Bollati et al. 2007; Pavanello et al. 2009; Pavanello et al. 2010; Yauk et al. 2008), and the effects of epigenetic changes on PAH-induced carcinogenesis warrants further investigation.

3. Bioactivation of PAH mixtures and reaction with DNA

In this section of the review the effects of interactions between PAHs in mixtures and the effects on bioactivation (i.e. effects on either expression or activity of metabolic enzymes like CYPs) and reaction with DNA (i.e. formation of adducts) is discussed. For this purpose we have sub-divided this section into two parts – interactions in binary PAH mixtures and those in complex PAH mixtures. To allow for comparisons of interaction effects we have mainly focused on studies that have co-exposed and compared effects of individual PAHs (such as B[a]P) with mixtures.

One aspect of the mechanism behind PAH-induced carcinogenesis that will not be discussed in this review is analysis of mutagenicity after exposure to complex PAH mixtures. Whilst this is a significant area of research for mixture effects, many of the studies have already been extensively reviewed in great detail previously (Claxton et al. 2004; Ohe et al. 2004; White 2002; White and Claxton 2004).

3.1. Effects of interactions in binary PAH mixtures

Early *in vivo* studies involved topical exposure of mice to binary PAH mixtures to investigate interaction effects on DNA binding (usually formation of B[a]P-DE-DNA adducts) and these are summarized in Table 1. Increased binding of B[a]P to DNA *in vivo* was the more prevalent observation following co-exposure of B[a]P with a number of PAHs including benzo[e]pyrene, dibenzo[a,e]pyrene, dibenzo[a,l]pyrene, fluoranthene or pyrene (Hughes and Phillips 1990; Rice et al. 1988; Rice et al. 1984; Slaga et al. 1979; Smolarek et al. 1987). Conversely, *in vivo* co-exposure of B[a]P with phenanthrene decreased DNA adduct formation (Rice et al. 1984), and with a mixture comprised of B[a]P with dibenzo[a,e]pyrene and dibenzo[a,l]pyrene, levels of DNA adducts were lower than observed for the individual PAHs (Hughes and Phillips 1990). Furthermore, topical co-

exposure of 7,12-dimethylbenz[a]anthracene with benzo[e]pyrene decreased binding of the former to mouse DNA (DiGiovanni et al. 1982; Slaga et al. 1979; Smolarek et al. 1987).

Whilst the above described studies did not investigate effects on bioactivation, the authors concluded that in the presence of different PAHs, activation leading to DNA binding intermediates is likely to be altered. In addition, it is also possible that the observed adduct levels in some studies could result from interactions affecting the activation of DNA repair systems. More recent in vitro and in vivo studies have investigated effects of binary PAH mixtures (Table 1). In in vitro studies using human-derived hepatocellular carcinoma (HepG2) cells exposed to binary PAH mixtures, increased levels of DNA adducts and expression of CYPs were generally observed compared to B[a]P alone (Cherng et al. 2001; Gabelova et al. 2013; Sevastyanova et al. 2007; Staal et al. 2007; Tarantini et al. 2011). Whilst the authors of these manuscripts attributed these responses to additive or more-than additive effects, only two of the manuscripts based these observations on different mathematical models of additivity (Gabelova et al. 2013; Staal et al. 2007) whereas the remaining observations were inferred from dose addition effects. Similar effects on CYP-mediated metabolism were observed in vivo in male Sprague-Dawley rats exposed to binary PAH mixtures (Bouchard et al. 1998) and increased DNA adduct levels have been observed in vitro in hamster embryo cells exposed to a mixture of B[a]P and benzo[e]pyrene (Smolarek and Baird 1984). A recent in vitro study in human-derived colorectal adenocarcinoma (HT-29) cells observed dose-dependent increased fluoranthene metabolism after co-exposure with B[a]P which the authors attributed to enhanced CYP-mediated activation of fluoranthene, although neither gene expression or CYP activity was measured directly (Harris et al. 2013).

Lower levels of DNA adducts and CYP expression after exposure to binary PAH mixtures have been observed *in vitro* in HepG2 and human embryonic lung (HEL) cells (Binkova and Sram 2004; Cherng et al. 2006; Sevastyanova et al. 2007), and whilst the responses in these studies are

consistent with less-than additive effects resulting from interactions, no appropriate mathematical calculation of additivity was included on which to confirm this. For the former it is likely that these contrasting findings in the HepG2 cells result from the use of high PAH concentrations (B[a]P 10 µM) and related effects on CYP inhibition as a result of elevated ROS formation (Cherng et al. 2006). For the latter this response probably results from the low metabolic capacity of HEL cells and the resulting decrease in PAH activation (Binkova and Sram 2004; Sevastyanova et al. 2007). Lower levels of DNA adducts and CYP expression/activity have in general been observed in vitro in rat cells (Falahatpisheh et al. 2004; Gabelova et al. 2013; Staal et al. 2008) and in vivo in killifish (Willett et al. 2001) and zebrafish (Van Tiem and Di Giulio 2011), and whilst the authors suggested these represented less than additive effects due to interactions, these conclusions were based on dose addition observations whereas only two studies applied mathematical modelling (Gabelova et al. 2013; Staal et al. 2008). It is possible that these contrasting findings result from differences between human- and experimental animal-derived cells and this conclusion is in agreement with the observed variations in sensitivity to TCDD (Silkworth et al. 2005). However, as the respective studies did not use the same test conditions (mixtures, exposure times, concentrations etc.) this conclusion regarding species sensitivity cannot fully account for the differences in response and it is probable that there are many different sources that contribute to the outcome.

An important caveat to consider for many of the above described studies on binary mixture effects *in vitro* is the used of the human HepG2 cell line. This particular cell line is one of the most extensively used for studying the effects of PAHs due to its demonstrated metabolic competency. However the result of this widespread use is that the particular cell line likely exists in many different clonal variants that might lack the inducibility of particular enzymes involved in bioactivation of PAHs. For example, it has been shown that hypermethylation of the *CYP1B1* gene promoter leads to a lack of induction of CYP1B1 after exposure to dioxin (Beedanagari et al. 2010) and this is another consideration that might explain the differences observed in the above described

studies. Furthermore this is a consideration that applies to all experimental cell lines and is an important reason for maintaining and using cells at low passage numbers to prevent accumulation of clonal differences.

In conclusion, the studies on binary PAH mixtures suggest that saturation of activation is an important consideration when studying interaction effects, and that the use of experimental models plays as important a role as the different mixtures studied. However, as there is a lack of consistency between the experimental conditions in the above described studies it is impossible to determine the driving source behind the observed differences. Since the described studies do not show a consistent effect in response to binary PAH mixtures on DNA damage levels, this suggests that formation of DNA damage is an inadequate marker for mixture effects.

3.2. Effects of interactions in complex PAH mixtures

To discuss interaction effects we have focused on studies where exposure to complex PAH mixtures has been investigated alongside exposure to individual PAHs. Studies that have investigated complex PAH mixtures have typically used two different types: either standard reference materials (SRM) or extracts obtained from collected environmental samples. SRMs from the National Institute of Standards and Technology have been used: from coal tar (CT, SRM1597, SRM1597a), from urban dust (UD, SRM 1649a), from diesel particulate extract (DPE, SRM1650b) and from diesel particulate matter (DPM, SRM1975) (May et al. 1992).

Two recently published reviews have extensively discussed transcriptional responses after exposure to complex PAH mixtures *in vitro* and *in vivo* from original studies and as such these effects have not been included (Huang 2013; Sen et al. 2007). Both reviews conclude that *in vitro* microarray

analysis patterns provide a comprehensive approach to assessing effects on gene expression that complement animal experiments.

The effects of exposure to SRMs both in vitro and in vivo are summarized in Table 2. Levels of DNA adducts in female SENCAR mice exposed in vivo to CT or UD with either B[a]P or dibenzo[a,l]pyrene decreased compared to the individual PAHs alone, although there was no clear correlation between adduct levels and change in CYP expression (Courter et al. 2007a; Marston et al. 2001). Similarly, lower levels of DNA adducts were observed in vitro in human-derived breast adenocarcinoma (MCF-7) cells and Chinese-hamster-derived lung fibroblast (V79) cells exposed to CT, UD or DPE with either B[a]P or dibenzo[a,l]pyrene (Mahadevan et al. 2005a; Mahadevan et al. 2005b; Mahadevan et al. 2007; Musafia-Jeknic et al. 2005). The decreases in DNA adduct levels are probably a result of increased competition for substrate binding to activating enzymes. In contrast, DNA adduct levels increased in female SENCAR mice exposed in vivo to DPE with either B[a]P or dibenzo[a,l]pyrene compared to the individual PAHs alone although again there was no clear correlation between adduct levels and change in CYP expression or activity (Courter et al. 2008). This difference in DNA adduct formation may be attributable to the higher amounts of DPE (50 mg) these mice were exposed to compared to the amount of either CT or UD (1 mg). It is also plausible that different PAHs in the DPE could be causing additive or more-than additive effects on DNA adduct formation, although in human-derived MCF-10A cells in vitro co-exposure of DPE with B[a]P caused decreased levels of DNA adducts compared to B[a]P alone, and co-exposure with dibenzo[a,l]pyrene caused no significant change (Courter et al. 2007b). Taken together these findings suggest that the response to complex PAH mixtures is likely attributable to inhibitory effects on activation resulting from particular PAHs in the mixture and potentially also the effects of metabolic enzyme saturation leading to less than additive effects although this was not mathematically modelled. Similar to the above described binary studies, these studies also suggest that DNA damage formation might not be a good marker for mixture effects.

To discuss the effects of interactions we have focused on studies that have co-exposed mixtures with individual PAHs and compared the effects. There is however a number of recent studies that have investigated the effects of exposure to complex PAH mixtures on DNA adduct formation compared to exposure to individual PAHs (without co-exposure). Exposure of human-derived BEAS-2B or MCF-7 cells or HEL cells in vitro to either SRMs or environmental PAH mixture extracts typically resulted in decreased levels of DNA adducts compared to B[a]P alone, although it is not always entirely clear the motivation for the choice of mixture concentration and it is plausible these decreases are reflective of the use of different concentrations (i.e. higher B[a]P when used alone than what is in the mixtures) (Binkova and Sram 2004; Kuljukka-Rabb et al. 2001; Oh et al. 2011; Pohjola et al. 2003; Rossner et al. 2013; Sevastyanova et al. 2007). Similarly, in vitro exposure of rat-derived WB-F344 cells or in vivo exposure of female FVB/N mice to complex PAH mixtures resulted in decreased levels of DNA adducts compared to B[a]P alone (Andrysik et al. 2011; Siddens et al. 2012). These findings suggest that interactions in complex PAH mixtures typically lead to less-than additive effects, similar to what was observed in the in vivo co-exposure experiments (Courter et al. 2008; Courter et al. 2007a; Marston et al. 2001). However, in in vitro studies using HepG2 cells increased DNA damage (both adducts and strand breaks) was observed after exposure to complex PAH mixtures suggesting additive/more-than additive effects of interactions (Niziolek-Kierecka et al. 2012; Sevastyanova et al. 2007; Tarantini et al. 2009). However, for both scenarios the assumptions of interaction effects are based on dose addition observations and were not determined mathematically. This is in agreement with the results obtained using binary PAH mixtures in HepG2 cells and probably reflects the metabolic competency of these cells (Knasmuller et al. 1998). Whilst interaction effects are not possible to determine and the proposed conclusions regarding additivity are based on simple dose addition observations, these studies show decreased levels of adducts with mixtures than with B[a]P which is in agreement with the above described co-exposure experiments indicating that analysis of B[a]P alone is not sufficient as a surrogate for mixture effects.

In conclusion, the observed effects resulting from exposure to complex PAH mixtures are similar to what was observed with binary PAH mixtures in that the response was dependent on many different contributory factors including the test system and mixtures assessed, and was not predictable from comparison to individual PAHs. However, similar to the conclusion from the binary studies the lack of coherence between the experimental conditions makes it impossible to determine the overall driving source of the responses.

4. Carcinogenicity of PAH mixtures

4.1. PAH exposure and cancer in humans

High exposures to PAHs in occupational settings are of significant risk to humans. Accordingly, IARC lists B[a]P and a number of industrial complex PAH mixtures including coal tars, mineral oils, shale oils and soot and exposure through cigarette smoke, aluminium production, coal gasification and coke production as carcinogenic to humans (IARC 2010). In addition, there are a number of individual PAHs and mixtures/exposures classified as groups 2A (probably carcinogenic) and 2B (possibly carcinogenic) (IARC 2010). Case-control studies on human workers exposed to PAHs through their occupation and also through dietary exposure have been extensively described previously (Boffetta et al. 1997; Bosetti et al. 2007; IARC 2010). From the case control studies human cancers associated with occupational exposure were observed in a large number of sites including the respiratory system, larynx, bladder, esophagus/stomach, skin and kidney. The most prevalent form of cancer from occupational exposure to complex PAH mixtures was found in the lungs (Boffetta et al. 1997; Bosetti et al. 2007; IARC 2010).

In addition to the above described occupational risk, a large number of factors have been linked to exposure to complex PAH mixtures and cancer. These include smoking, cooking and dietary intake and wood burning. Tobacco smoking is considered the overwhelming cause of lung cancer in humans and is associated with approximately 90 % of all lung cancer incidences and 1.2 million deaths per year worldwide (DeMarini 2004; IARC 2004). In addition, smoking has been identified as a cause of cancer in a number of other human organ sites (IARC 2004). B[a]P was one of the first carcinogenic compounds detected in tobacco smoke, and the links between tobacco smoking, B[a]P and DNA damage have been reviewed recently (Alexandrov et al. 2010). Dietary exposure to complex PAH mixtures is also a significant risk for human cancer development (Phillips 1999), and moreover, is the primary route of non-occupational exposure for non-smokers. A number of human cancers have been associated with dietary intake of complex PAH mixtures including colorectal, pancreatic and prostate (IARC 2010). In addition to food consumption, cooking has also been linked to development of human cancer. Investigations in the county of Xuan Wei, China, showed unusually high levels of lung cancer mortality rates. In this area, residents (particularly women) are exposed to high levels of unvented smoke from indoor cooking resulting in high level and chronic exposure to complex PAH mixtures which was highly associated with lung cancer development (Chuang et al. 1992; DeMarini et al. 2001). More work is required to fully understand the cellular mechanisms involved in carcinogenesis but it is evident that exposure to PAHs in all these scenarios is a major contributor to the observed cancers.

4.2. Carcinogenic effects of PAH mixtures in experimental animals

In the two-stage model tumors can be induced by application of an initiator (carcinogenic PAH, sub-threshold dose) followed by repetitive exposure to a promoter (Mason et al. 1990; Nesnow et al. 1983). Over 100 PAHs and their metabolites are known to be tumor initiators or complete

carcinogens in the mouse skin two-stage model (Nesnow et al. 1983; Pereira 1982) with the consequence that interactions most likely alter the response to complex PAH mixtures compared to individual PAHs.

Studies on binary PAH mixtures have shown that weak or non-carcinogenic PAHs have various influences on the tumor-initiating activities of PAHs with stronger carcinogenic potential (Table 1). Dermal exposure of mice to B[a]P with various PAHs caused an increase in tumor initiation which correlated with increased DNA binding. Conversely, dermal co-exposure of 7,12dimethylbenz[a,h]anthracene with the same PAHs caused a decrease in tumor initiation which correlated with decreased 7.12-dimethylbenz[a,h]anthracene binding to DNA (DiGiovanni et al. 1982; Slaga et al. 1979). It was suggested that the co-exposed PAHs might be affecting the metabolism to DNA-reactive metabolites. In a dose-response study performed in mice it was found that quintary PAH mixtures exhibited more than additive effects at low doses and less than additive effects at high doses when lung adenomas were scored compared to individual PAH exposure (Nesnow et al. 1998). This paradoxical finding (i.e. non-linear dose-effect relationship) probably results from competitive inhibition of the metabolizing enzymes and hence decreased amounts of DNA-reactive metabolites and fits with the findings of the above described studies on binary and complex PAH mixtures.

Tumor induction resulting from exposure to complex PAH mixtures has been studied *in vivo* using a number of different exposure routes including oral (Culp et al. 1998; Goldstein et al. 1998; Weyand et al. 1995), gavage (Robinson et al. 1987), dermal (Deutsch-Wenzel et al. 1984; Grimmer et al. 1985; Grimmer et al. 1983; Grimmer et al. 1982; Nesnow et al. 1983) and lung implantation (Grimmer et al. 1988; Grimmer et al. 1987a; Grimmer et al. 1987b; Grimmer et al. 1984). A review by Schneider et al. (2002) of some of these earlier tumor induction studies concluded that the contribution of B[a]P to the overall potency of the PAH mixtures depends on the route of exposure

and the end-point studied (i.e. the cancer type) (Schneider et al. 2002). More recent studies using the dermal tumour initiation mouse model and co-exposure of complex PAH mixtures with either B[a]P or dibenzo[a,l]pyrene are summarized in Table 3 with the reported effects on numbers of tumors per tumor bearing animal (TBA) as an indicator of response (Courter et al. 2008; Courter et al. 2007a; Marston et al. 2001). One important conclusion was that the formation of DNA adducts in the skin (after single administration) did not predict the final tumor response suggesting that DNA adduct formation is a poor marker for tumorigenesis induced by complex PAH mixtures.

In these studies when the complex PAH mixtures were co-exposed with either B[a]P or dibenzo[a,l]pyrene increased numbers of tumour/TBA were observed compared to the mixture alone, whereas no effect or decreased numbers of tumors/TBA were observed when compared to exposure to B[a]P or dibenzo[a,l]pyrene alone (Table 3). Taken at face value these findings seem to suggest that the effects on tumors/TBA are solely governed by the application of the individual PAHs. In the study published by Siddens et al. (2012) these discrepancies were discussed in the context of RPF values and how using values from the U.S. EPA the RPF value for a mixture of DPM and CT would be 0.34 μ g B[a]P-equivalent ($B[a]P_{eq}$). In comparison to B[a]P, which was used at 100 μ g, it would be predicted that the mixtures would elicit a much weaker tumor response. However, their results showed no differences in incidence, latency, multiplicity or tumor type between the two exposures showing that the observed tumor effects in response to the complex mixture were higher than anticipated based on current RPF values (Siddens et al. 2012). It should however be considered that not all PAHs in the mixtures tested would have been assigned RPF values due to the lack of experimental data which could lead to misestimating of the mixture potency.

Using a similar approach we calculated the corresponding $B[a]P_{eq}$ values of the CT, DPE and UD mixtures used in three recent mice studies (Courter et al. 2008; Courter et al. 2007a; Marston et al.

2001). The RPF values were 0.23, 0.28 and 0.015 μ g B[a]P_{eq} respectively, to be compared with the 50.4 μ g B[a]P used and in stark difference to the observed numbers of tumors/TBA which differed around a factor of 3. These results support the observation from Siddens et al. (2012) that much weaker responses than were observed would be predicted, and highlights the inadequacy of the current RPF system for complex PAH mixtures which is further discussed below.

5. Risk assessment of complex PAH mixtures

5.1. Risk assessment approaches for PAHs

Cancer risk assessment of PAHs typically encompasses three approaches -B[a]P as a surrogate marker for mixture exposure, the use of component-based factors (such as RPF and TEF values) or the comparative potency approach (Flowers et al. 2002; IPCS 1998; Pufulete et al. 2004). The first approach has been applied to determine risk of exposure to PAHs in air (WHO 2000) and by the European Food Safety Agency (EFSA) as a marker for PAH levels in foods (EFSA 2008). EFSA further investigated the use of mixtures of other PAHs (PAH4 and PAH8) and concluded these were more appropriate markers of PAH exposure in foodstuffs (EFSA 2008). However, the necessary assumption that relative PAH concentrations will remain stable between different exposure scenarios is a major weakness for ubiquitous PAH mixtures and both agencies concluded that the surrogate marker approach is likely to misestimate the actual risk of exposure to humans. In the second approach the different components of the PAH mixtures (where identifiable) are assigned potency factors relative to B[a]P (i.e. RPF or TEF values) and from these factors cancer risk estimates for PAH mixtures can be derived. Risk estimates are determined by calculating and summing B[a]P_{eq} values for the different PAHs in the mixtures, and as described above, this can be limited by a lack of potency values for all PAHs. To adjust for possible adverse effects resulting from interactions uncertainty factors (or safety factors) are often included in risk assessments (ECETOC 2003; ECETOC 2010; KEMI 2003; Martin et al. 2013). The final approach assesses the whole PAH mixture by estimating the potency without having to identify or quantify individual PAH compounds. Using a source-specific PAH mixture which has been thoroughly characterized in epidemiological and biological studies as a reference, a second sufficiently similar PAH mixture is ranked based on its comparative potency in the biological test. This approach has been suggested for a variety of complex PAH mixtures (Albert et al. 1983; Lewtas 1985), although to the authors' knowledge, has yet to be implemented by regulatory agencies. In the following section the limitations of these approaches to complex PAH mixture risk assessment are discussed and then an alternative approach based on whole mixture assessment is presented.

5.2. Limitations of the current approaches for PAH mixture risk assessment

Risk assessment of complex PAH mixtures is a challenging issue. As discussed throughout this review there is a growing consensus that the current approaches to risk assessment are insufficient. The weaknesses of these approaches regarding complex PAH mixture assessment are outlined below.

- *Ubiquitous nature of complex PAH mixtures*: The composition of complex PAH mixtures in the environment varies significantly depending on location sampled, time/season of sampling and surrounding industrial and vehicular use. This means that analysis of a surrogate PAH (i.e. B[a]P) for whole mixtures is unreliable when predicting human risk. An option to restrict this limitation has been suggested involving identification of typical PAH profiles or "fingerprints" correlated to the different sources (Howsam and Jones 1998; Ravindra et al. 2008; Tobiszewski and Namieśnik 2012).
- Lack of a common mechanism for PAHs: For the component-based approach (i.e. using RPF or TEF values) to be valid for complex PAH mixture risk assessment it requires a common mode of action and observed additivity in the resulting toxicological effect. The

heterogeneous mechanisms that are likely to occur for different PAHs (Boström et al. 2002) predispose that this is not the case, and the many studies described in this review demonstrating non-additive effects (both more- and less-than) arising from interactions confirm this.

Disparity between TEF/RPF values for individual PAHs: For the majority of PAHs data regarding their potency is lacking and therefore many PAHs have not been assigned potency values. A range of RPF or TEF values from 8 different published scales (Collins et al. 1998; EPA 2004; Kroese et al. 2001; Larsen and Larsen 1998; Malcolm and Dobson 1994; Muller 1997; Nisbet and LaGoy 1992; U.S.EPA 2010) is given in Table 4. It is evident from the ranges that there are significant variations in the assigned factors depending on which test systems they are based on that will be reflected when calculating B[a]Peq values for mixtures and is likely to misestimate the risk, in agreement with earlier conclusions about the limitations of equivalency factors (Gaylor et al. 2000; Schneider et al. 2002). Dibenzo[a,I]pyrene is a very good example of this as depending on the scale used the factor can vary between 1 and 100 (Table 4) and has recently been suggested to be even higher (Siddens et al. 2012). Similarly, dibenz[a,h]anthracene ranges from 0.1 - 10 suggesting that it is either ten-fold more or less potent than B[a]P (Table 4). In many cases these differences can be attributed to the derivation of values from different experimental exposures and cancer studies. However, it remains that the lack of coherence between the different scales is a major limitation for risk assessment.

One of the biggest issues for risk assessment of mixtures of PAHs is the dearth of information available. This lack of information has been highlighted by both the EU and the U.S. EPA as being a significant issue (EC 2009; U.S.EPA 2000). A prioritization of data needs and collection of reliable and relevant data for risk assessment is critical to overcome this issue and a number of

considerations have been discussed by these agencies previously. Some of the most important considerations are outlined below:

- The need to identify and characterize whole mixtures more extensively and the adoption of procedures to model various human exposure scenarios. Better characterization of the individual PAHs in mixtures is important for understanding their contribution to the mixture and assignment of potency values to more PAHs in mixtures would improve component-based risk assessment strategies (although this should not predispose risk assessment away from whole mixture analysis).
- Shift towards assessment of whole mixture effects and exposure, including chemicals with dissimilar structures and modes of action. More empirical evidence of the joint action of environmental mixtures would pre-dispose towards more sensitive and accurate risk assessment and likely challenge the rule of thumb assumption that dose addition is sufficient for predicting risk.
- Improvement of the dose-response methods for modelling risk including development of stringent methods for validating these. These models should be robust to include all interaction effects and not just dose addition.
- In many cases mixtures of PAHs have been chosen to suit certain experimental conditions leading to a trend in studying mixtures with known/similar effects and a lack of information on chemicals that alone do not induce effects but can act to modulate toxicity. Future assessment would be better suited by including these chemicals
- Current risk assessment methods tend towards a single risk estimate. A priority towards risk
 estimate based on exposure ranges would be more appropriate, ideally reflecting different
 risk analysis methods for the particular mixtures and thus allowing for incorporation of
 uncertainties associated with method choice.

Although the component based approach is recommended and mostly used by both the US and EU authorities, the implementation of a whole mixture based approach is necessary in order to account for possible interaction effects. This has previously been recognized by the U.S. EPA (Flowers et al. 2002; U.S.EPA 2000) and more recently by the US National Toxicology Program (NTP) which has approved a research program aiming to increase the knowledge of complex PAH mixture toxicity (NTP 2012). This approach has so far not been a viable option mainly because of limited data for establishing reference mixtures and methods for determining sufficient similarity.

A potential alternative would be to use mixture assessment factors (MAFs) similar to those that have been discussed previously (Backhaus et al. 2010). In a report for the Swedish Chemicals Agency an approach was discussed outlining how a MAF could be used to reflect a default exposure scenario through single substance assessment for situations where mixtures are present with neither a defined chemical composition (i.e. where it is impossible to ascertain the different components and respective amounts) and where quantitative knowledge regarding individual component toxic potencies is lacking (Backhaus et al. 2010). MAFs, similar to the comparative potency approach, differ to component-based approaches in that they do not rely on individually assigned potency values compared to a reference compound but instead compare the effects of whole mixtures on a relevant biological endpoint. However, in contrast to the comparative potency approach, the MAFs do not rely on a well characterized and sufficiently similar reference mixture but instead on using a well characterized single PAH (i.e. B[a]P).

We have recently investigated the potential of using activation of DNA damage signaling as a biological end-point for developing MAFs (Jarvis et al. 2013; Niziolek-Kierecka et al. 2012). DNA damage signaling governs the detection of damage and downstream activation of cell cycle arrest, DNA repair and/or cell death and thereby plays an important role in the response to carcinogens

(Figure 2). Using DNA damage signaling for studying mixtures also has the advantage that whilst the different types of DNA damage induced by PAHs have different kinetics of formation, activate different repair processes, and demonstrate different efficiencies for repair, they share a common DNA damage signaling network directing all of these processes (Ciccia and Elledge 2010; Sancar et al. 2004).

DNA damage signaling results from activation of Ataxia telangiectasia mutated (ATM), ataxia telangiectasia and Rad3-related protein (ATR) or DNA-dependent protein kinase (DNA-PK). Activation of ATM and DNA-PK is commonly attributed to DNA double strand breaks whereas activation of ATR results from single stranded DNA (Lovejov and Cortez 2009). Two proteins that are activated downstream (by phosphorylation) in the response to the different types of damage are Checkpoint kinase 1 (Chk1), a kinase involved in cell cycle regulation (Zeng et al. 1998; Zhao and Piwnica-Worms 2001) and H2A histone family member X (H2AX), a core histone protein maintaining genomic stability (Fernandez-Capetillo et al. 2004; Rogakou et al. 1998). Phosphorylation of H2AX is an established marker for DNA damage (Fernandez-Capetillo et al. 2004; Mah et al. 2010) and recent studies of the role of phosphorylated Chk1 in the DNA damage signaling network have confirmed its potential as a marker (Dai and Grant 2010; Ma et al. 2011; Patil et al. 2013). Both proteins have been shown as markers of DNA damage in vivo (Bauer et al. 2011; Bauer et al. 2012) and furthermore have been shown to be activated in response to single and double strand breaks, bulky DNA adducts and oxidative DNA damage, all of which are induced by PAHs as described above (Bihari and Fafandel 2004; Mattsson et al. 2009; Park et al. 2005; Park et al. 2006; Rogan et al. 1993; Szeliga and Dipple 1998; Tung et al. 2014). It is considered that the different types of DNA damage could give rise to different patterns of activation of DNA damage signaling and hence different response proteins. Thus, it is likely necessary that for more accurate risk assessment analysis of more than one marker might be required to cover all the different possible patterns of activation.

Recently it has been demonstrated that both Chk1 and H2AX are efficiently phosphorylated in response to DNA damage following exposure to low levels of environmental mixtures of PAHs (Jarvis et al. 2013; Mattsson et al. 2009; Niziolek-Kierecka et al. 2012). In a study exposing HepG2 cells to nanomolar concentrations of PAHs extracted from urban air PM persistent phosphorylation of Chk1 and H2AX was observed at B[a]P_{eq} concentrations of 1 and 8 nM respectively (Jarvis et al. 2013) showing that these markers are sensitive for low levels of DNA damage. Using B[a]P as a reference compound the results suggested that the mixture was at least 100 times more potent at B[a]P_{eq} concentrations than exposure to B[a]P alone (Jarvis et al. 2013). This is in line with the *in vivo* studies discussed above (Courter et al. 2008; Courter et al. 2007a; Marston et al. 2001; Siddens et al. 2012) and suggests that phosphorylation of Chk1 and H2AX could serve as biological endpoints for developing MAFs. Further work is required to investigate the effects of exposure to low-levels of DNA damaging compounds and whether the induction of these markers correlates with results obtained from animal studies.

5.4. Limitations of the current approaches for PAH mixture risk assessment

Risk assessment based on whole mixtures is proposed to be more appropriate in future for determining the risks of exposure to PAH mixtures on human health. However, there are a number of issues that have to be considered as outlined below:

- It is impossible to test all mixtures in animal studies. From both an ethical and a practical perspective it is not possible to perform animal studies for every mixture. Analysis of the effects of similar mixtures is a potential way to avoid this issue but that approach is not viable for scenarios where mixture composition changes.
- Challenge to determine what the mixture will be compared against. As environmental mixtures are comprised of many different chemical classes it is a challenge to determine

how the whole mixtures will be compared experimentally. For example, comparison to B[a]P is only sufficient for the PAH component of the mixture. Furthermore a reliable biological endpoint is required for whole mixture assessment and, although we discuss above how DNA damage signalling is a possibility, it remains to be determined what marker(s) would be most appropriate.

- Chemical characterization is costly and difficult. Whilst it is not fundamentally needed for all whole mixture based risk assessment strategies the characterization of the chemical mixtures for use as a reference mixture (for example in the comparative potency approach) is difficult for a number of reasons including lack of analytical information on individual PAHs for characterization and overlapping of molecular signatures making it hard to analyze low-level PAHs or PAHs with increasing ring numbers.
- Whole mixture assessment does not permit experimental analysis of chemical modes of action. For the purpose of risk assessment this aspect is arguably less important than the above issues. However, important for experimental research is a need to understand the mode of action of chemicals and this is impossible to determine in mixtures, only the mode of action of the mixture is testable.

6. Conclusions and future challenges for complex PAH mixture risk assessment

Determining the health risks to humans of exposure to complex PAH mixtures is profoundly challenging and a daunting task. In this review we have described the current status of the field of mixture effects with regards to PAHs *in vitro* and *in vivo*, and why the current approaches for risk assessment are scientifically limited and likely to misestimate the actual risk to human health.

One of the biggest issues for understanding complex PAH mixture toxicity is that it is impossible to test all mixtures in animal models and cultured cell lines. It is widely accepted that both approaches

have limitations that restrict their extrapolation to human exposures, but at present they remain the best tools for experimental research. It is evident from the studies described in this review that the observed responses to complex PAH mixtures are dependent on the modeling system used. Similar to what the *Toxicity Testing in the 21st Century* program promotes (NRC 2007; Schmidt 2009) and NTP has recently given approval for (NTP 2012), in the future a combination of different approaches to evaluating complex PAH mixture effects should be used alongside animal models and cultured cell lines to better understand the effects of interactions. This combination could include other approaches such as tissue models generated in bioreactors and human induced pluripotent stem cells, and also *in silico* modeling, although much more work is required to understand how these can be successfully integrated into risk assessment.

In addition to robust modeling systems, it is necessary to identify reliable, sensitive and relevant markers for risk assessment of complex PAH mixtures which importantly are quantifiable in different experimental settings. The analysis of PAH activation (by CYPs) and study of DNA adduct formation have demonstrated wide variation in the response to complex PAH mixtures (see above described studies) showing they are poor markers for mixture effects. At this point it remains unclear what other markers would be appropriate, though the inclusion of analysis of activation of DNA damage signaling would appear to be compatible.

Finally, the emphasis needs to be switched to studying real life mixtures. It is clear from the above described studies that binary PAH mixture assessment, whilst allowing for general experimental analysis of effects, represents a poor surrogate for studying mixture effects in relation to human risks. An important change for the future and one which has taken precedent in more recent analyses is to investigate the effects of complex PAH mixtures, ideally mixtures that are extracted from actual environmental samples such as urban air and soil. It is important that risk assessment focuses on whole mixtures allowing for the effects of interactions to be included. A greater

identification of the different chemicals in the mixtures would be beneficial for understanding deposition and human exposure, but should not predispose risk assessment towards component-based analyses.

In conclusion, it is apparent that there are significant challenges to overcome to implement reliable and accurate risk assessment of complex PAH mixtures. Paramount to this is implementation of analyzing relevant mixtures that reflect actual human exposure. It is important to consider that today no one single approach is reliable or acceptable and that a combination of different techniques and strategies is required. Finally, a concerted effort on behalf of the researchers and the regulatory agencies is required to develop more intelligent and systematic approaches to mixture-based risk assessment.

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Figure Legends

Figure 1. Examples of some commonly studied PAHs (A) and substituted PAHs (B). The bay and fjord regions are indicated by arrows.

Figure 2. DNA damage signalling coordinates the genotoxic insults caused by complex PAH mixtures. A large variety of DNA damages including strand breaks, adduct formation, oxidative damage and abasic sites resulting from depurinating damages occur following exposure to complex PAH mixtures. Central to the response to these damage stimuli is DNA damage signalling. Tolerable DNA damage can lead to downstream responses including cell cycle arrest and DNA repair, whereas excessive/irreparable DNA damage can lead to apoptosis. Both scenarios are considered positive responses as they prevent the carcinogenic process. Mutations can occur as a result of unrepaired or wrongly repaired DNA damage or errors in signal transduction and are a negative outcome as they promote and accelerate the carcinogenic process. The pivotal role that DNA damage signalling plays in coordinating the DNA damage response makes it an appropriate biological endpoint for whole mixture assessment.

Table 1: Summary of the effects on CYP450s and DNA adduct formation after in vivo and in vitro exposure to binary PAH mixtures compared to individual PAH effects

T 4 C 4	DATING	ъ	Dant		CYP450s			D 6	
Test System	PAH Mixture	Dose	Route	Studied	EROD	Effect	- Adduct Levels	Reference	
		In vivo	studies						
Female CD-1 Mice	B[a]P:B[e]P/PYR/FLU	200°:100°	Topical				+	Slaga et al. 1979	
	DMBA:B[e]P	10 ^a :100 ^b	Topical				-	_	
	B[a]P:FLU/PYR	15-22 ^a :171-183 ^a	Topical				+	Rice et al. 1984	
	B[a]P:FLU	11 ^a :110 ^a	Topical				+	Rice et al. 1988	
Female SENCAR Mice	B[a]P:B[e]P	200:20-200 ^a	Topical				=	DiGiovanni et al. 1982	
	DMBA:B[e]P	20:20-200 ^a	Topical				+		
Tomale BEIVET III WHEE	B[a]P:B[e]P	50-200:50-2000 ^a	Topical				+	Smolarek et al. 1987	
	DMBA:B[e]P	5-20:5-400 ^a	Topical				-		
Male Parkes Mice	DB[a,l]P:B[a]P/DB[a,e]P	500:500 ^a	Topical				=	Hughes and Phillips 1990	
	B[a]P:DB[a,e]P	500:500 ^a	Topical				+		
	DB[a,l]P:B[a]P:DB[a,e]P (tertiary mixture)	500:500:500 ^a	Topical				-		
Male Sprague-Dawley Rats	PYR:B[a]P	5:0.5-25°	i.v.		No	+ (met)		Bouchard et al. 1998	
	PYR:NAP	5:0.5-25 ^c	i.v.		No	+ (met)			
Fundulus heteroclitus	B[a]P:FLU	5:5-50 ^d	i.p.	1A	Yes	- (act)	+, -	Willett et al. 2001	
Danio rerio	B[k]F:FLU	50:150 ^e		1A, 1B1, 1C1	Yes	+ (exp), - (act)		Van Tiem and Di Giulio 2011	
		In vitro	studies						
	B[a]P:B[ghi]P	1:0.1-10 ^f		1A1	No	+ (exp)	+	Cherng et al. 2001	
	B[a]P:1-NP	10:2.5-10 ^f		1A1	No	- (exp)	-	Cherng et al. 2006	
	B[a]P:B[b]F/DB[a,h]A/DB[a,l]P/1-MP	3:3/1/0.1/30 ^f		1A1/2, 2B6	No	+ (exp)	+	Staal et al. 2007	
	B[a]P:FLU	3:30 ^f					+	Staar et al. 2007	
HepG2 cells (human)	B[a]P:B[a]A/B[b]F/B[ghi]P/CHR/DB[a,h]A/IP	1:1 ^f					+	Sevastyanova et al. 2007	
riepoz cens (numan)	B[a]P:B[k]F	1:1 ^f					-	Sevastyanova et al. 2007	
	B[a]P:B[b]F/DB[a,h]A/IP	1:1 ^f					+		
	B[a]P:B[k]F	1:1 ^f					-	Tarantini et al. 2011	
	B[a]P:B[a]A/B[ghi]P/CHR/FLU/PYR	1:1 ^f					=		
	B[a]P:DB[c,g]C	0.01-1:1 ^f		1A1	No	+ (exp)	+,-	Gabelova et al. 2013	
	B[a]P:B[a]A/B[b]F/B[k]F/B[ghi]P/CHR/DB[a,h]A/IP	1:1 ^f					-	Binkova and Sram 2004	
Embryonic lung cells (human)		0.1:0.1 ^f					-	Blikova alid Statil 2004	
	B[a]P:B[a]A/B[b]F/B[k]F/B[ghi]P/CHR/DB[a,h]A/IP	1:1 ^f					-	Sevastyanova et al. 2007	
HT-29 cell (human)	FLU:B[a]P	1-25:1-25 ^f			No	+ (met)		Harris et al. 2013	
Rat kidney cells	B[a]P:CHR	0.03-30:3 ^f		1A1, 1B1	Yes	- (exp), - (act)		Ealahatriahah at al. 2004	
	B[a]P:ANT	0.03-30:3 ^f		1A1, 1B1	Yes	+ (exp), - (act)		Falahatpisheh et al. 2004	
WB-F344 cells (rat)	B[a]P:DB[c,g]C	0.01-1:1 ^f	 	1A1	No	- (exp)	-	Gabelova et al. 2013	
	B[a]P:B[b]F/DB[a,h]A/DB[a,l]P	3:10/10/0.3 ^f		1A2	No	- (exp)	-		
Precision-cut rat liver slices	B[a]P:FLU	$3:30^{f}$		1A2	No	+ (exp)	+	Staal et al. 2008	
	DB[a,h]A:DB[a,l]P/FLU	10:0.3/30 ^f		1A2	No	- (exp)	-		
	All mixtures			1A1	No	+ (exp)			
Hamster embryo cells	B[e]P:B[a]P	0-5a:11.2-18.9b					+	Smolarek and Baird 1984	

Abbreviations: 1-MP, 1-methylphenanthrene; 1-NP, 1-nitropyrene; act, EROD activity; ANT, anthracene; B[a]A, benz[a]anthracene; B[a]P, benzo[a]pyrene; B[b]F, benzo[b]fluoranthene; B[e]P, benzo[e]pyrene; B[ghi]P, benzo[ghi]perylene; B[k]f, benzo[k]fluoranthene; CHR, chrysene; DB[a,e]P, dibenzo[a,e]pyrene; DB[a,h]A, dibenz[a,h]anthracene; DB[a,l]P, dibenzo[a,l]pyrene; DB[c,g]C, dibenzo[c,g]carbazole; DMBA, 7,12-dimethylbenz[a]anthracene; exp, gene expression; FLU, fluoranthene; i.p., intraperitoneal; i.v., intravenous; IP, indeno[123-cd]pyrene; met, metabolism; NAP, naphthalene; PYR, pyrene

Symbol guide: a, nmol; b, µg; c, µmol/kg; d, mg/kg; e, µg/L; f, µM; +, increase compared to PAH alone; -, decrease compared to PAH alone; =, no observed difference in effect compared to PAH alone;

Table 2: Summary of the effects on CYP450s and DNA adduct formation after in vivo and in vitro co-exposure to complex PAH mixtures compared to individual PAH effects

Test System	PAH Mixture	Dose	Route		CYP	450s	- Adduct Levels	Reference	
Test System				Studied	EROD	Effect	Adduct Levels		
				In vivo stu	dies				
Female SENCAR Mice	B[a]P:CT	200a:1b	Dermal	1A1, 1B1	No	+ (exp)	-	Marston et al. 2001	
	DB[a,l]P:CT	2a:1b	Dermal	1A1, 1B1	No	+ (exp)	-	Maiston et al. 2001	
	B[a]P:UD	200 ^a :1 ^b	Dermal	1A1, 1B1	No	- (exp)	-	Courter et al. 2007a	
	DB[a,l]P:UD	200a:1b	Dermal	1A1, 1B1	No	- (exp)	-		
	B[a]P:DPE	200 ^a :50 ^b	Dermal	1A1, 1B1	No	- (exp), - (act)	+	Courter et al. 2008	
	DB[a,l]P:DPE	2a:50b	Dermal	1A1, 1B1	No	+ (exp), = (act)	+		
				In vitro stu	ıdies				
	B[a]P:CT	80°:400°		1A1, 1B1	No	= (exp)	-	M-h-d	
	DB[a,l]P:CT	$0.8^{a}:400^{c}$		1A1, 1B1	No	+ (exp)	-	Mahadevan et al. 2005b	
MCE 7 C II (II	B[a]P:UD	80°:400°		1A1, 1B1	Yes	+ (exp), - (act)		Mahadevan et al. 2005a	
MCF-7 Cells (Human)	DB[a,l]P:UD	$0.8^{a}:400^{c}$		1A1, 1B1	Yes	+ (exp), - (act)	=		
	B[a]P:UD	10:400°		1A1, 1B1	Yes	+ (exp), = (act)		Musafa-Jeknic et al. 2005	
	DB[a,l]P:UD	$0.1:400^{\circ}$		1A1, 1B1	Yes	+ (exp), + (act)	-		
MCF-10A Cells (Human)	B[a]P:DPE	80°:400°		1A1, 1B1	Yes	= (exp), = (act)		Courter et al. 2007b	
	DB[a,l]P:DPE	$0.8^{a}:400^{c}$		1A1, 1B1	Yes	+/= (exp), +/= (act)	=		
V70 C-11- (Chinasa III-matan)	B[a]P:CT	80°:400°					-	M-11	
V79 Cells (Chinese Hamster)	DB[a,l]P:CT	$0.8^{a}:400^{c}$					-	Mahadevan et al. 2007	

Abbreviations: B[a]P, benzo[a]pyrene; CSC, cigarette smoke condensate; CT, coal tar; DB[a,l]P, dibenzo[a,l]pyrene; DPE, diesel particulate extract; DPM, diesel particulate matter; UD, urban dust Symbol guide: a, nmol; b, mg; c, µg; d, mgSRMeq/mlCrudeExtract; +, increase compared to PAH alone; -, decrease compared to PAH alone; =, no observed difference in effect compared to PAH alone





